**Density Notes**

All forms of \_\_\_\_\_\_\_\_\_\_\_\_\_ have \_\_\_\_\_\_\_\_\_\_\_\_\_\_ and \_\_\_\_\_\_\_\_\_\_\_\_\_ properties.

Physical properties include:

* + \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
  + \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
  + \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
  + \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

**Density**: A measure of how much \_\_\_\_\_\_\_\_\_\_ is contained in a given \_\_\_\_\_\_\_\_\_\_\_\_.

**Formula**…..

**Simplified Formula**

**We LOVE density**

Why is density important?

* Density is a \_\_\_\_\_\_\_\_\_\_\_\_\_ that can be used to \_\_\_\_\_\_\_\_\_\_\_\_\_ a substance.
* Density tells whether an \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
  + An object \_\_\_\_\_\_\_\_\_\_\_\_\_ if it is \_\_\_\_\_\_\_\_\_ dense than the liquid \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.

Density is made up of two measurements

* \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ & \_\_\_\_\_\_\_\_\_\_\_\_\_\_
* Density is expressed as a \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_:
* grams per cubic centimeter \_\_\_\_\_\_\_\_\_\_ and grams per milliliter \_\_\_\_\_\_\_\_\_\_
* Remember \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

**Water**

* Water’s density is \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

**Practice Problems**

1. What is the density of a wood block with a volume of 100 cm3 and a mass of 25 g?
2. What is the density of a liquid with a mass of 45 g and a volume of 48 mL?
3. A cube has a mass of 2.8g and occupies a volume of 3.67mL. Would this object float or sink in water? Why?
4. A liquid has a mass of 25.6 g and a volume of 31.6 mL. Calculate its density.

What is the identity of the liquid? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

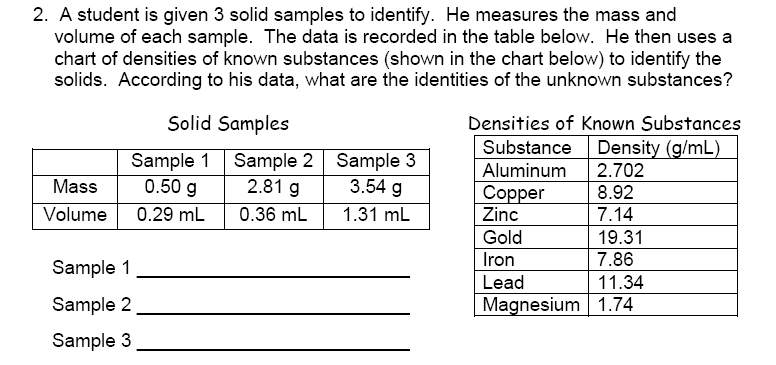
\*Use the information below for reference.

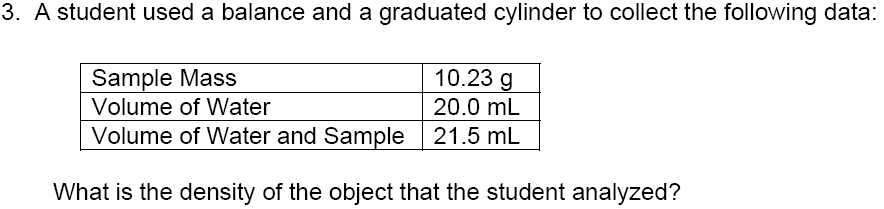
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| --- | --- |
| **Substance** | **Density (g/mL)** |
| Mercury | 13.6 |
| Water | 1.0 |
| Ethanol | 0.81 |

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5.

6.





7.